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(54) PRODUCTION OF HYDROGEN PEROXIDE

(57) Abstract:

PROBLEM TO BE SOLVED: To produce H2O2 high in conc. under low pressure with high hydrogen selectivity by using a specified catalyst when H2O2 is produced by catalytically reacting O2 with H2 in a reaction medium.

SOLUTION: The catalyst used is prepared by depositing titania and platinum group metal (e.g. Pd) on a carrier preferably comprising silica or alumina, or by depositing titania on a carrier and further depositing

platinum group metal. Titania is preferably obtd. by hydrolyzing titanium alkoxide (e.g. ortho ethylitanate) as the source material to obtain titanium hydroxide, depositing the obtd. titanium hydroxide on the carrier surface, and calcining preferably at 300 to 1,000°C. H₂O₂ is produced by catelytically reacting H₂ with O₂ in a reaction medium which is a soln, containing acid (e.g. phosphoric acid) and halide (NaBr) in the presence of the catalyst at the pressure of 0.1 to 2MPa.

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